

0040-4039(94)01613-5

ASYMMETRIC SYNTHESIS OF THE ANTIBIOTIC (+)-THIAMPHENICOL USING CIS-N-(p-TOLUENESULFINYL)AZIRIDINE 2-CARBOXYLIC ACIDS

Franklin A. Davis* and Ping Zhou Department of Chemistry, Drexel University, Philadelphia, PA 19104

Summary: A concise, highly efficient asymmetric synthesis of aminopropanediol (1R,2R)-(-)-3, precursor to the broad spectrum antibiotics thiamphenicol/florfenicol 1/2, was prepared in two steps from cis-aziridine 2-carboxylic acid (2S,3S)-(-)-5.

The broad-spectrum, antibacterial, synthetic antibiotic thiamphenicol, (1R,2R)-(+)-2-(dichloroacetamido)-1-[(4-methylsulfonyl)phenyl]-1,3-propanediol (1), is active against many grampositive and gram-negative microorganisms.¹ Its fluoro analog, florfenicol, (-)-2, exhibits much higher activity with a toxicity comparable to 1.² The current process for manufacturing thiamphenicol (+)-1,^{1b} and hence florfenicol (-)-2,³ requires a classical resolution of (±)-*threo*-2amino-1-[(4-methylthio)phenyl]-1,3-propanediol (3) which in turn is prepared in a multi-step sequence.^{4,5} While only the (1*R*,2*R*)-(-)-3 isomer is a useful precursor to 1 and 2, three procedures have been developed for transforming the inactive (1*S*,2*S*)-(+)-3 isomer into (-)-3. Two of these methods involve a direct conversion of (+)-3 into (-)-3 via a series of steps^{6,7} while the third entails a racemization which ultimately results in a second resolution.⁸ Enzymatic resolution of a derivative of (±)-3 has also been reported to give aminopropanediol (-)-3 in 36% yield.⁷ In this context we describe a concise, highly stereoselective asymmetric synthesis of the key thiamphenicol/florfenicol precursor (1*R*,2*R*)-(-)-3 from enantiopure (2*S*,3*S*)-(-)-N-(*p*-toluenesulfinyl)-2-carbomethoxy-3-(4methylthiophenyl)aziridine (5) and its conversion to (+)-1.



Synthesis of aziridine (-)-5 was accomplished using our recently reported Darzens-type synthesis of cis-N-(*p*-toluenesulfinyl)aziridine 2-carboxylic acids, precursors of the difficult to prepare syn- or *threo*- β -hydroxy- α -amino acid structural unit.⁹ Thus, addition of a 3.0 mmol THF solution of (*S*)-(-)-N-(4-methylthiobenzylidene)-*p*-toluenesulfinimine (4)¹⁰ at -78 °C to 7.5 mmol of the lithium enolate of methyl bromoacetate, prepared in the usual way from lithium bis(trimethylsilyl)amide, at -78 °C gave (-)-5 in 55-60% yield following flash chromatography (Scheme).¹² Importantly the trans isomer was not detected in the crude reaction mixture.

Scheme

OMe 3 eq. LiAlH₄ 78 °C (55-60%) (87%) (25,35)-(-)-5 (S)- (-)-4 TsOH/H₂O Cl₂CHCOCI/Et₃N (93%) (94%) ŌН (1R,2R)-(-)-3 (25,35)-(+)-6 2.5 eq m-CPBA (1R,2R)-(+)-1 (88%) ŌН (1R,2R)-(+)-7

Ar= *p*-MeSPh

Treatment of aziridine (-)-5 with 3.0 equiv. of lithium aluminum hydride (LiAlH₄) in ether at 0 °C, warming to rt for 1 h and quenching with sat. NaHCO₃ solution gave (1H)-(2S,3S)-(+)-3-[(4-methylthio)phenyl]aziridine-2-methanol (6) in 87% isolated yield.¹³ That the aziridine ring remained intact was confirmed by the similarity in J_{2,3}, of 5 and 6, 7.3 vs 7.2 Hz, respectively and its conversion into (-)-3. Not only was the carbomethoxy group in 5 reduced, but deprotection of the N-*p*-toluenesulfinyl group also occurred. To the best of our knowledge this represents the only example of the reductive removal of an N-sulfinyl group.^{14,15}

Ring-opening of aziridines requires activation at nitrogen, and only a few studies have been reported for aziridines unsubstituted at nitrogen.^{16,17} For N-unsubstituted 2-carboxylic acids, Lewis acid or acid activation is often sufficient for reaction to occur. However, if the ring contains an activating substituent such as phenyl or *p*-methoxyphenyl opening is not always stereospecific, indicating that the transition state for attack at C3 has considerable carbocation character.^{9,16a} In the synthesis of (±)-sphinganine (*erythro*-2-amino-1,3-dihydroxyoctadecane), ring hydrolysis resulted in a 70:30 mixture of isomers and necessitated refluxing the cis-aziridinium hydrochloride in benzene with Amberlyst A 26.^{17b} Under similar conditions, however, the more reactive trans 3-pentadec-(1E)-enyl derivative gave (±)-*erythro* sphingosine as a single isomer.^{17c}

For hydrolytic ring-opening of **6** we chose *p*-toluenesulfonic acid (TsOH) because of its solubility in organic solvents and low nucleophilicity. Stirring **6** (0.25 mmol) with 1.1 equivalents of TsOH in a 1:1 water/THF mixture at rt for 30 min., followed by removal of the solvent, dilution with water and bring the solution to pH 12 with 50% NaOH gave, on extraction into CH_2Cl_2 , (-)-3 in 93% isolated yield.⁷ The reaction was exceptionally clean and remarkably stereospecific affording this key thiamphenicol/florfenicol precursor as a single isomer despite the presence of the activating 3-*p*-methylthiophenyl group in the aziridine.

Transformation of 2-amino-1,3-propanediol (-)-3 into thiamphenicol (+)-1 was readily accomplished by first forming the N-dichloroacetamide and oxidation of the sulfide to the sulfone. Treatment of (-)-3 with a slight excess of dichloroacetyl chloride and triethylamine afforded (+)-7^{1b} in 94% yield and oxidation with 2.6 equivalents of 95% *m*-CPBA in THF and quenching with sat. Na₂S₂O₃ gave the antibiotic in 88 % isolated yield.^{1b,3}

A concise, highly efficient asymmetric synthesis of the key thiamphenicol/florfenicol precursor (1R,2R)-3 has been prepared in three steps, in 49% overall yield, from readily available enantiometric pure sulfinimine (S)-4. The procedure avoids the wasteful resolution of racemic 3 used in earlier procedures.

Acknowledgments: This work was support by the National Science Foundation and the National Institutes of Health (GM 34014).

REFERENCES AND NOTES

- (a) Elks, J.; Ganellin, C. R. *Dictionary of Drugs*, Chapman and Hall: London, 1990, T-00179.
 (b) Cutler, R. A.; Stenger, R. J.; Suter, C. M. *J. Am. Chem. Soc.* 1952, *74*, 5475.
- (a) Elks, J.; Ganellin, C. R. Dictionary of Drugs, Chapman and Hall: London, 1990, F-00124.
 (b) Nagabhushan, T. L. U.S. Patent, 4235892, 1980, Schering Corp.; Chem. Abstr. 1980, 94, 139433.
- 3. For leading reference see: Schumacher, D. P.; Clark, J. E.; Murphy, B. L.; Fischer, P. A. J. Org. Chem. 1990, 55, 5291.

- Long, L. M. U. S. Patent, 2767213, 1956; Chem. Abstr. 1957, 51, 7414b. Collet, A.; Brienne, M. J.; Jacques, J. Bull. Soc. Chim. Fr. 1972, 127. Jacquez, J.; Collet, A.; Wilen, S. in Enantiomers, Racemates and Resolutions, John Wiley and Sons, New York, 1981, p 223.
- 5. Tyson, R. Chem. Ind. 1988, 118.
- 6. Giordano, C.; Cavicchioli, S.; Levi, S.; Villa, M. J. Org. Chem. 1991, 56, 6114.
- 7. Clark, J. E.; Fischer, P. A.; Schumacher, D. P. Synthesis, 1991, 891.
- 8. Giordano, C.; Cavicchioli, S.; Levi, S.; Villa, M. *Tetrahedron Lett.* **1988**, *43*, 5561. Horak, V.; Moezie, F.; Klein, R. F. X.; Giordano, C. *Synthesis* **1984**, 839.
- 9. Davis, F. A.; Zhou, P.; Reddy, G. V. J. Org. Chem. 1994, 59, 3243.
- Sulfinimine (-)-4 was prepared in 80% yield as previously described¹¹ from commercially available (1*R*,2*S*,5*R*)-(-)-menthyl (*S*)-*p*-toluenesulfinate and *p*-(methylthio)benzaldehye and had the following properties: mp 132-4 °C; [α]_D²⁰ -40.2° (c 1.1, CHCl₃); ¹H NMR (CDCl₃) δ 2.39 (s, 3H), 2.50 (s, 3H), 7.25 (d, J=8.3Hz, 2H), 7.30 (d, J=8.3Hz, 2H), 7.62 (d, J=8.3Hz, 2H), 7.72 (d, J=8.3Hz, 2H,), 8.68 (s, 1H).
- 11. Davis, F. A., Reddy, R. E. Szewczyk, J. M.; Portonovo, P. Tetrahedron Lett. 1993, 34, 6229.
- 12. Aziridine (-)-5 had the following properties: mp 86-8 °C; $[\alpha]_D^{20}$ -2.3° (c 0.5, CHCl₃); ¹H NMR (CDCl₃) δ 2.42 (s, 3H), 2.47 (s, 3H), 3.41 (s, 3H), 3.47 (d, J=7.3Hz, 1H), 3.82 (d, J=7.3Hz, 1H), 7.20 (d, J=8.2 Hz, 2H), 7.33 (d, J=8.1Hz, 2H), 7.39 (d, J=8.2Hz, 2H), 7.70 (d, J=8.1Hz, 2H).
- 13. Aziridine-2-methanol (+)-6 had the following properties: mp 125-125 °C; $[\alpha]_D^{20}$ +96.8° (c, 0.7, CHCl₃); ¹H NMR (CDCl₃) δ 1.84 (br, 2H), 2.48 (s, 3H), 2.64 (AB_q, J= 6.4Hz, 1H), 3.25 (dd, J₁=7.2 Hz, J₂ = 11.8 Hz, 1H), 3.44 (m, 2H), 7.24 (q_{AB}, J=8.6Hz, 4H).
- 14. Reduction of sulfinimines to sulfinamide with LAH has been reported. See: Annunziata, R.; Cinquini, M.; Cozzi, F. J. Chem. Soc. Perkin I, **1982**, 339.
- For deprotection of N-sulfinyl group under acidic conditions, see: Cinquini, M.; Cozzi, F. J. Chem. Soc. Chem. Commun. 1977, 723. Hua, D. H.; Miao, S. W.; Chen, J. S.; Iguchi, S. J. Org. Chem. 1991, 56, 4. Davis, F. A.; ThimmaReddy, R.; Reddy, R. E. J. Org. Chem. 1992, 57, 6387. See also reference 14.
- For leading references to the ring-opening reactions of aziridines see: a) Legters, J.; Thijs, L.
 Zwanenburg, B. *Recl. Trav. Chim. Pays-Bas*, **1992**, *111*, 16. b) Legters, J.; Willems, J. G. H.;
 Thijs, L.; Zwanenburg, B. *Recl. Trav. Chim. Pays-Bas*, **1992**, *111*, 59.
- For references to ring-opening of aziridine 2-methanols under acidic conditions, see: a) Bongini, A.; Cardillo, G.; Orena, M.; Sandri, S.; Tomasini, C. J. Chem. Soc. Perkin Trans I, 1986, 1339. b) Bongini, A.; Cardillo, G.; Orena, M.; Sandri, S.; Tomasini, C. J. Chem. Soc. Perkin Trans I, 1986, 1345. c) Cardillo, G.; Orena, M.; Sandri, S.; Tomasini, C. Tetrahedron, 1986, 42, 917.

(Received in USA 15 July 1994; accepted 18 August 1994)